

ICYS 2023 DFT Crash Course

Practice on Al metal

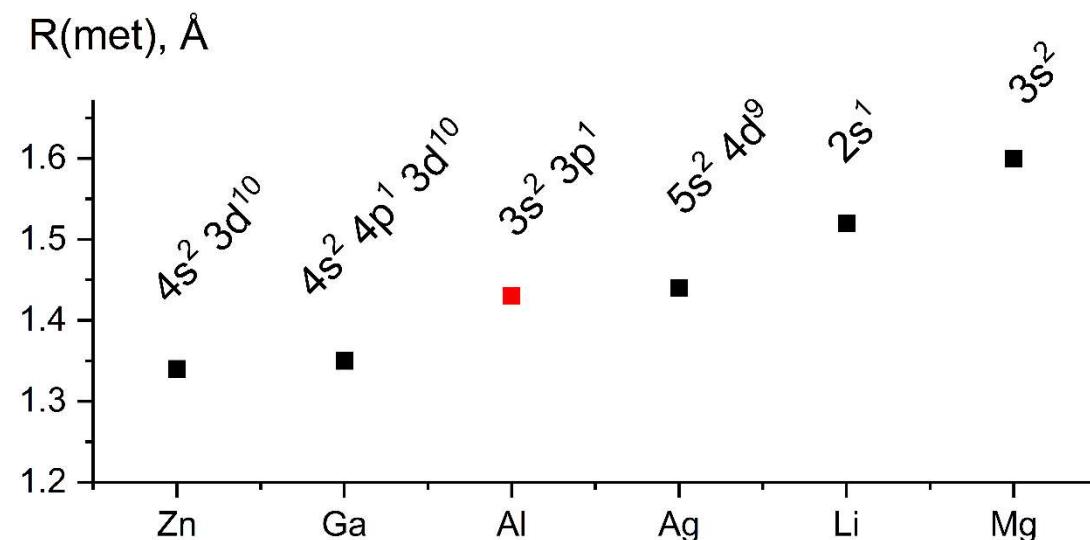
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12 ‘friends’ of Al metal

Compositions of solid solutions, at.-%:

Zn	65.5	Si	1.5
Ag	23.8	Mn	0.71
Li	16.2	Cr	0.37
Mg	18.9	V	0.2
Ga	5.5	Sc	0.35
Cu	2.5	Ti	0.146

!!!
**Maximum
compositions up to
500 °C**



We take only 6 of them

Mg, Zn – hcp

Al, Ag – ccp

Li – bcc

Ga – distorted hcp (Cmce)

Calculation set

VASP with ‘simanrc’ module

Standart optimization set: (PAW, PBE)

- EDIFF = 1e-05
- NSW = 20
- EDIFFG = -0.025
- IBRION = 1
- ISIF = 3
- ENCUT = 300
- ENAUG = 700.0
- KSPACING = 0.2
- LREAL = Auto
- ISMEAR = 0



Changed on ISIF = 4 for hexagonal Mg and Zn

Changed on ENCUT = 150 for comparison

+

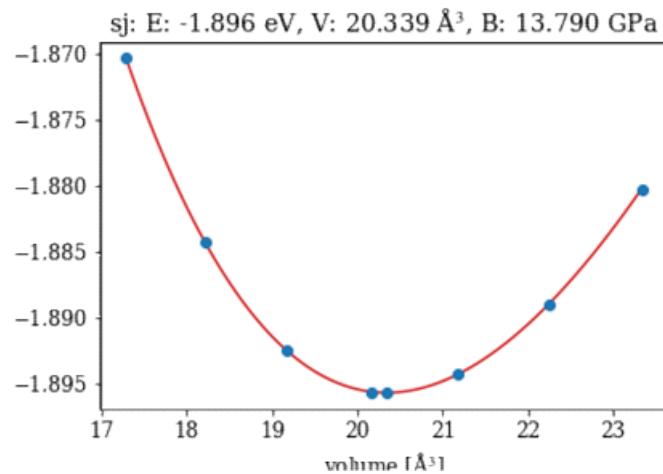
2 supercells for defects:

fcc-cell => 4 atoms per cell

- 32 atoms => 8 cells [2*2*2]
- 108 atoms => 27 cells [3*3*3]

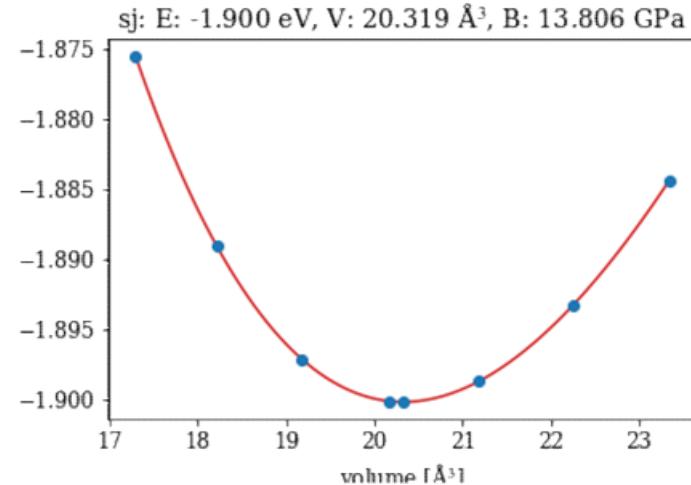
Metals optimization

ENCUT 150

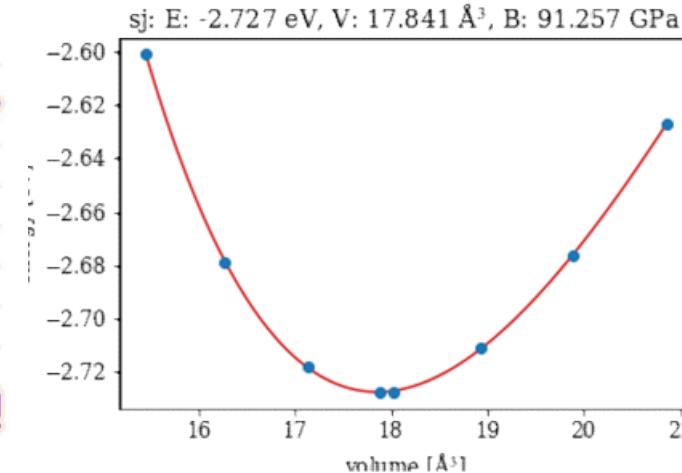
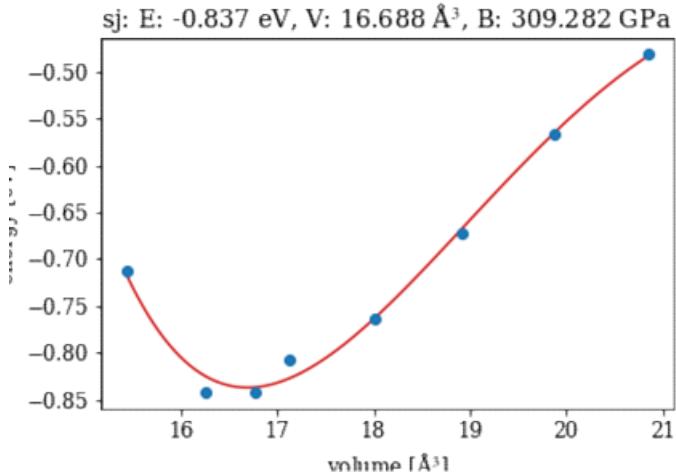


Li

ENCUT 300



Ag

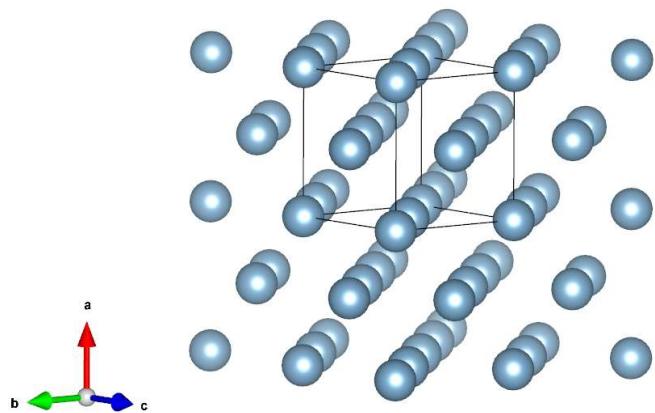


Vacancy

$$E_v(\text{Experimental}) = 0.66 \text{ eV}$$

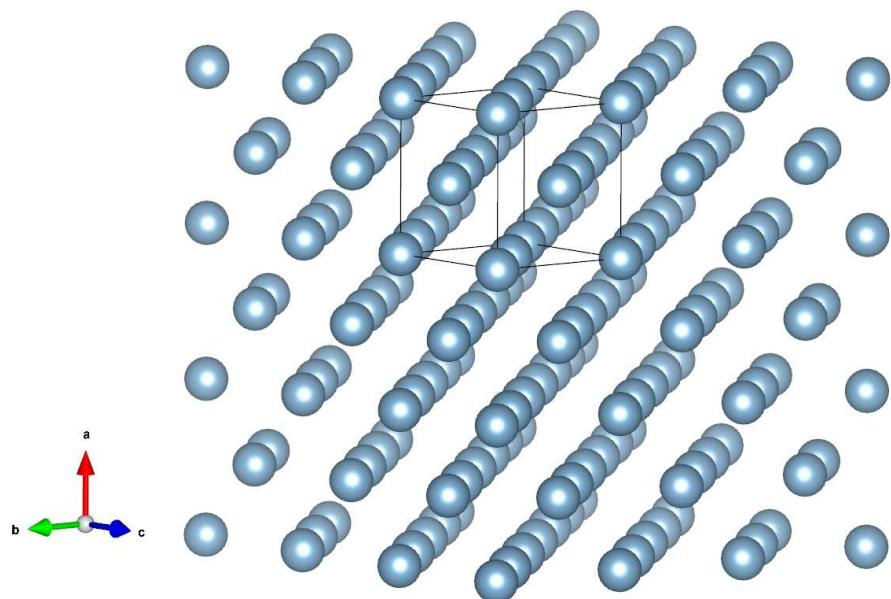
$$[2^*2^*2] - E_v = 0.69 \text{ eV}$$

$$1/32 = 3.1\%$$



$$[3^*3^*3] - E_v = 0.63 \text{ eV}$$

$$1/108 = 0.9\%$$



$$V_{\text{vac}} = 33 \text{ \AA}^3$$

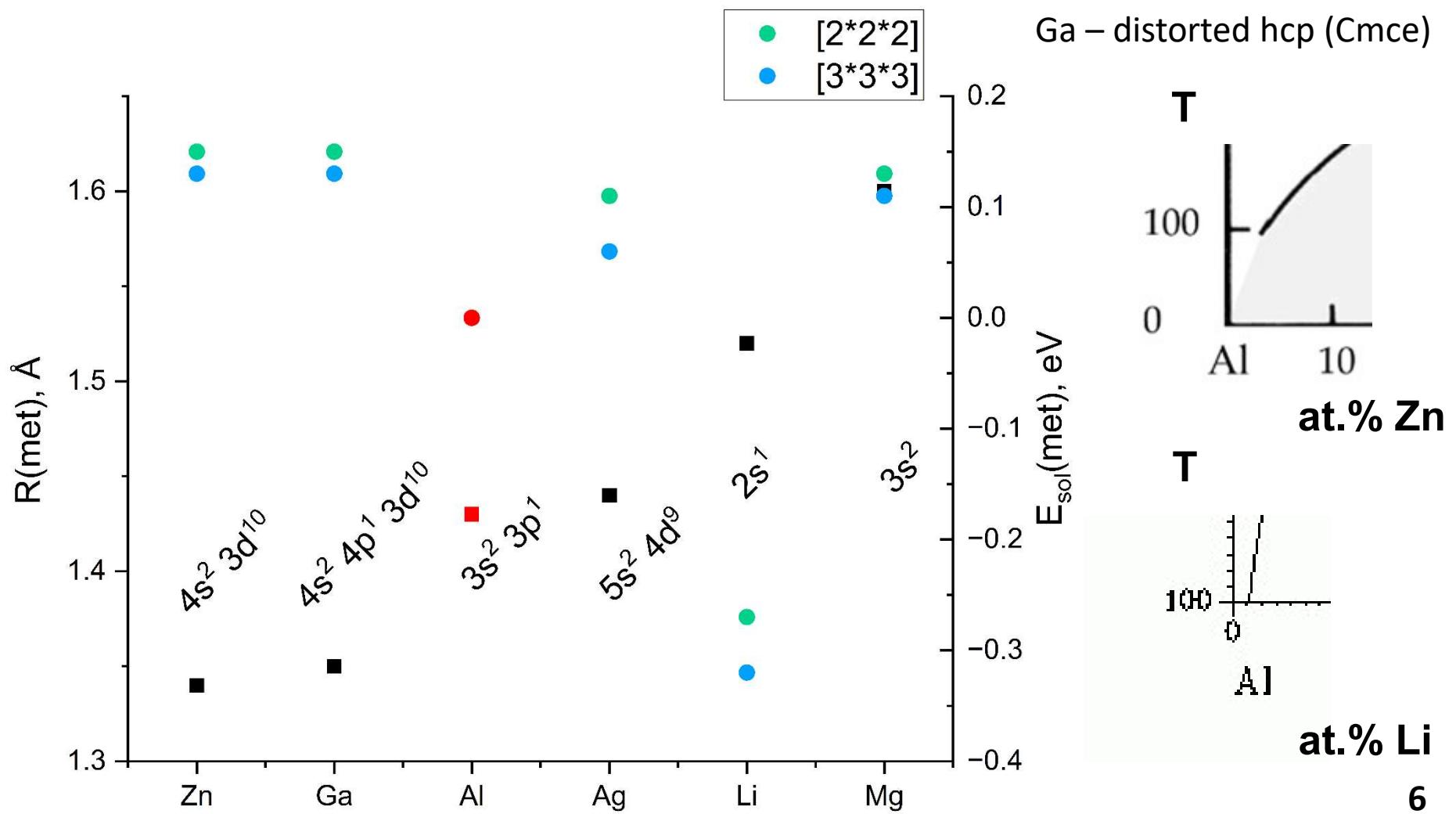
$$V_{\text{sphere}} \sim 4r_{\text{met}}^3$$

$$r_{\text{met}} = 1.3 - 1.6 \text{ \AA} \Rightarrow V_{\text{at}} = 8.7 - 16.4 \text{ \AA}^3$$

Vacancy

$$V_{\text{vac}} = 33 \text{ \AA}^3$$

$$V_{\text{sphere}} \sim 4r_{\text{met}}^3 \text{ but } T = 0 \text{ K}$$



Binding energy

$$V_{\text{vac}} = 33 \text{ \AA}^3$$

$$V_{\text{sphere}} \sim 4r_{\text{met}}^3 \text{ but } T = 0 \text{ K}$$

Mg, Zn – hcp

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